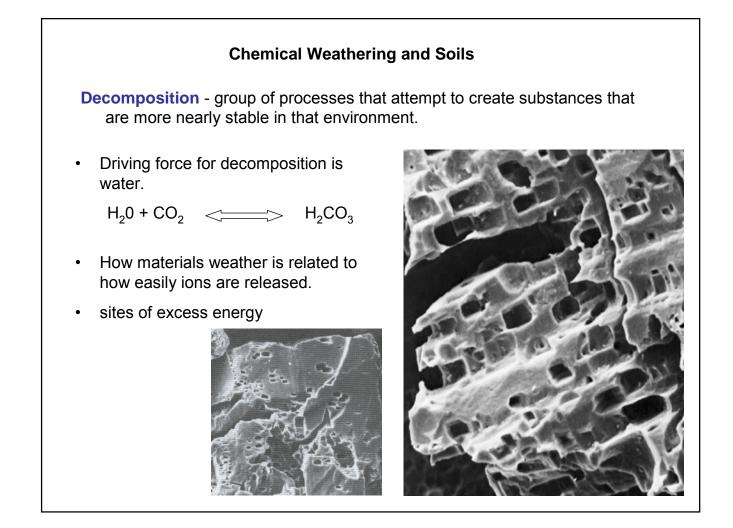


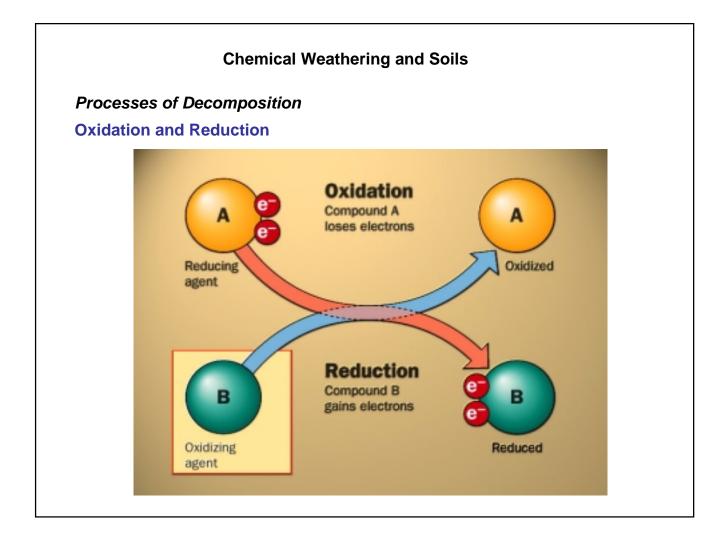
- 1. new minerals created by the weathering processes,
- 2. minerals that resisted destruction, and
- 3. organic debris added to the weathering zone.



Functions of weathering

- gives rock lower strength and greater permeability,
- produces minor landforms,
- releases minerals in solution,
- Is first step in soil formation.

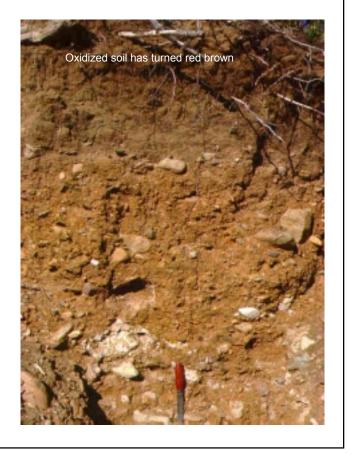




Processes of Decomposition

Oxidation and Reduction

 iron is the most commonly oxidized mineral element Fe⁺² (ferrous iron) —> Fe⁺³ (ferric iron) or 2FeO + O₂ —> Fe₂O₃

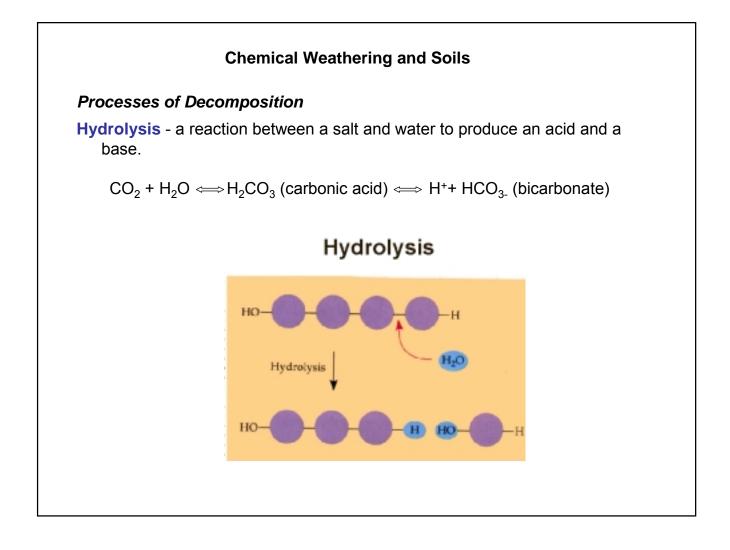


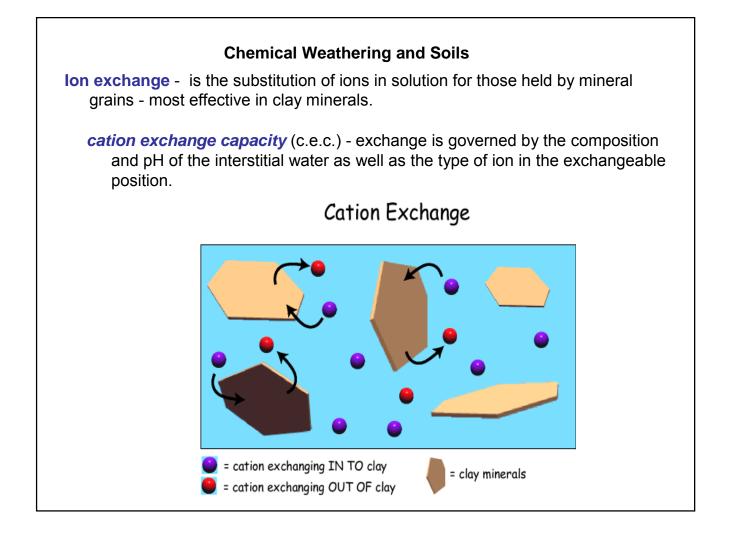
Processes of Decomposition

Solution - when atoms are dissolved from a mineral, the structure becomes unstable.

 $CaCO_3 + H_2CO_3 \longrightarrow Ca^{+2} + 2HCO_{3-}$

 bicarbonate represents the largest constituent of the dissolved load of most rivers





Representative Cation Exchange Capacities for Various Materials			
Material	Approximate cation exchange capacity (me/100 g dry weight)		
Organic matter	150-500		
Kaolinite	3-15		
Halloysite	5 - 10		
Hydrated halloysite	40-50		
Illite	10 - 40		
Chlorite	10-40		
Smectite	80-150		
Vermiculite	100 - 150 +		
Palygorskite	5-30		
Sepiolite	20-45		
Allophane	25-70		
Hydrous oxides of aluminum and iron	4		
Feldspars	1-2		
Quartz	1-2		
Basalt	1-3		
Zeolites	230-620		

(Taken from Carroll 15 and Grim $^{32}\!\!;$ palygorskite and sepiolite data from Weaver and Pollard. $^{71}\!\!)$

10

Mobility of decomposed materials

Leaching

- removes in solution the constituents that have been separated from minerals
- affects the pH of fluids surrounding minerals
- transfer mechanism

Mobility of decomposed materials

Fixation and retardation - mobility is considerably lower than might be expected.

Chelation - reaction between a metal ion and a complexing agent

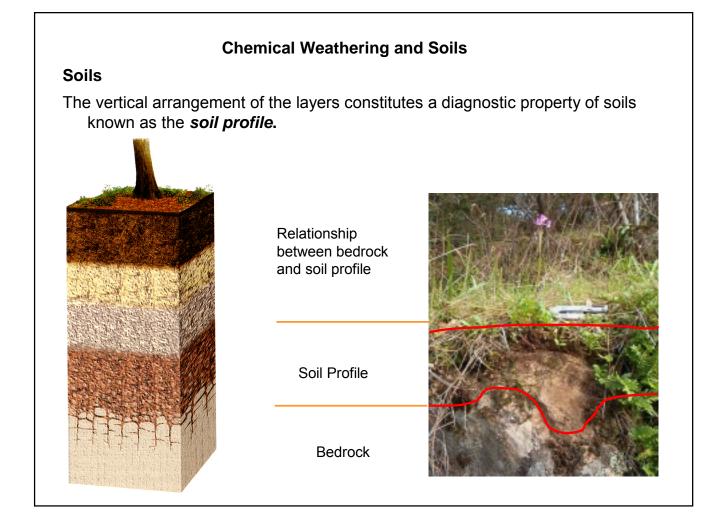
• metallic ions that are extremely immobile under normal conditions can be mobilized by reacting with complexing agents

Degree and rate of Decomposition

- End products
- Mineral stability
- Secondary minerals
- Estimates based on chemical analyses

Mean Lifetime of a 1 mm Crystal at 25°C and pH 5 (From Lasaga et al., 1994, Table 1 and References Therein).

Mineral	al Lifetime (years)	
Quartz	34,000,000	
Kaolinite	6,000,000	
Muscovite	2,600,000	
Epidote	923,000	
Microcline	921,000	
Prehnite	579,000	
Albite	575,000	
Sanidine	291,000	
Gibbsite	276,000	
Enstatite	10,100	
Diopside	6,800	
Forsterite	2,300	
Vepheline	211	
Anorthite	112	
Wollastonite	79	

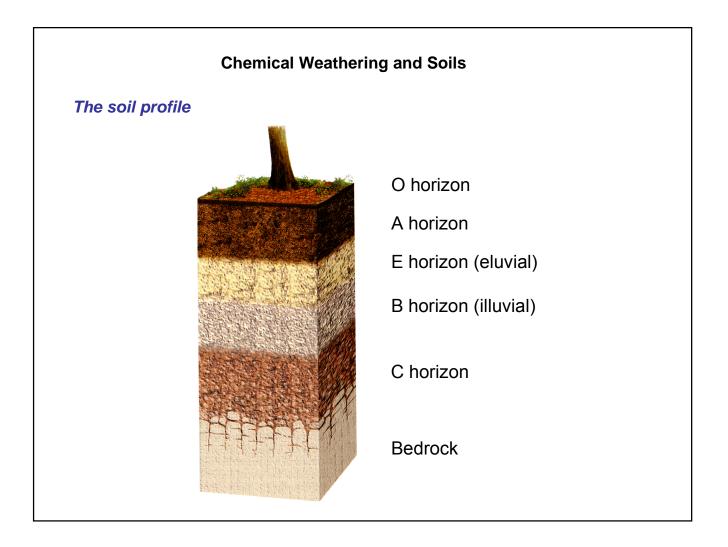


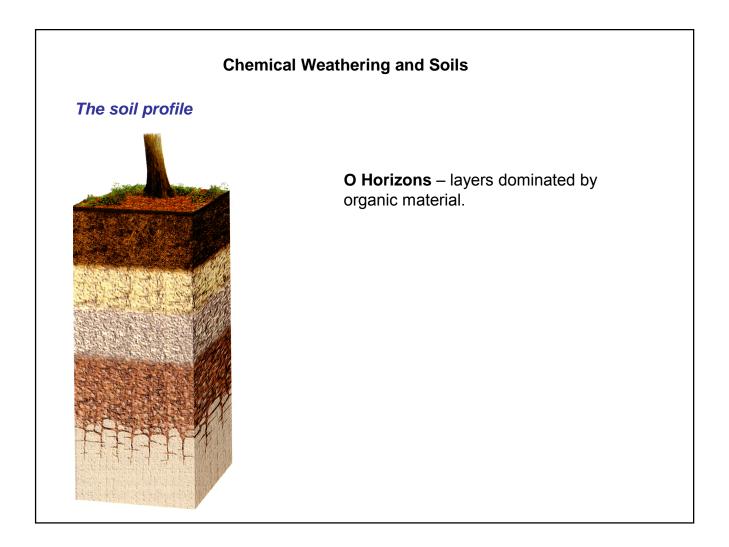
Fundamental concepts of soil genesis (pedogenesis)

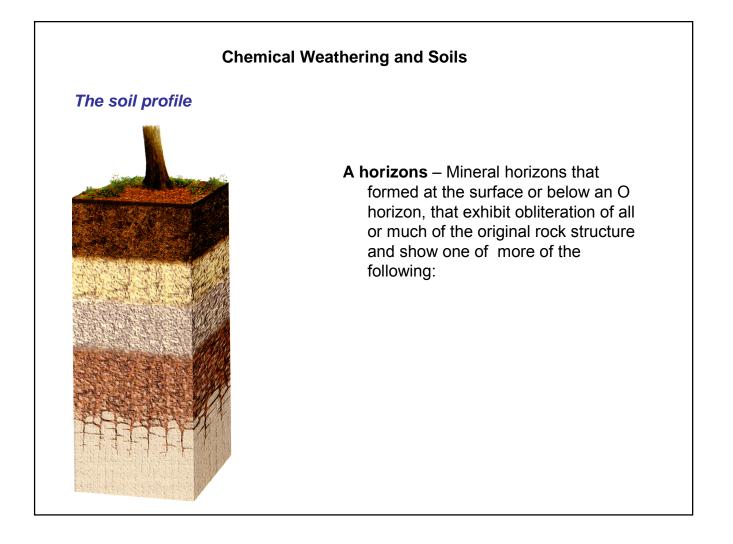
- Present day processes operate through space and time,
- Distinct regimes of soil processes produce distinct soils,
- Soil and its vegetation cover modify processes of land degradation,
- Clay is produced within the soil,
- Organo-mineral complexes are produced in the soil,

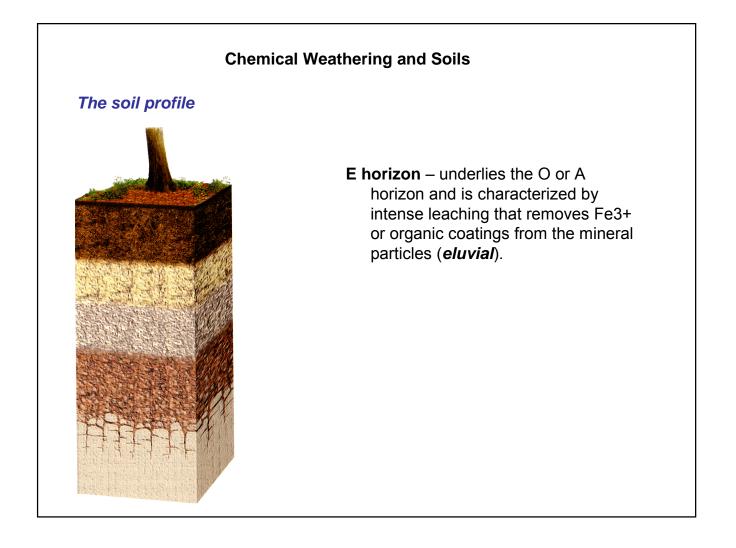
Fundamental concepts of soil genesis (pedogenesis)

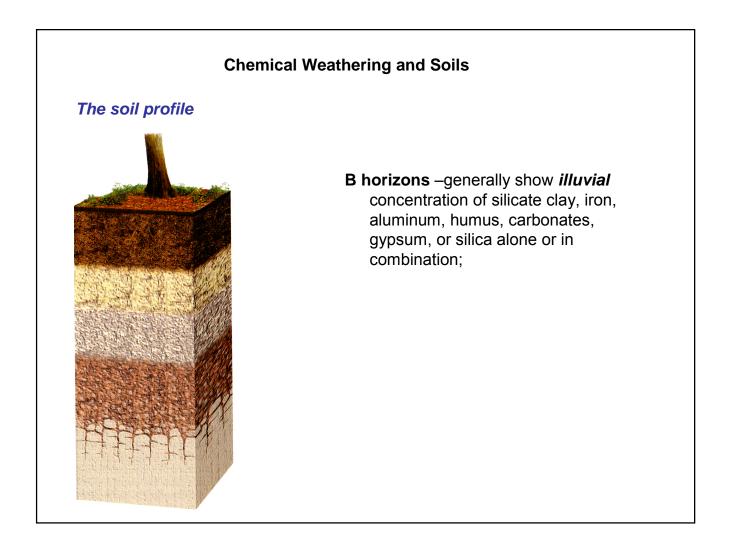
- In the course of pedogenesis, soil succession occurs,
- Complexity is more common than simplicity
- Little of the soil continuum is older than Tertiary and most no older than Pleistocene times,
- A knowledge of the Pleistocene is pre-requisite to understanding soils.

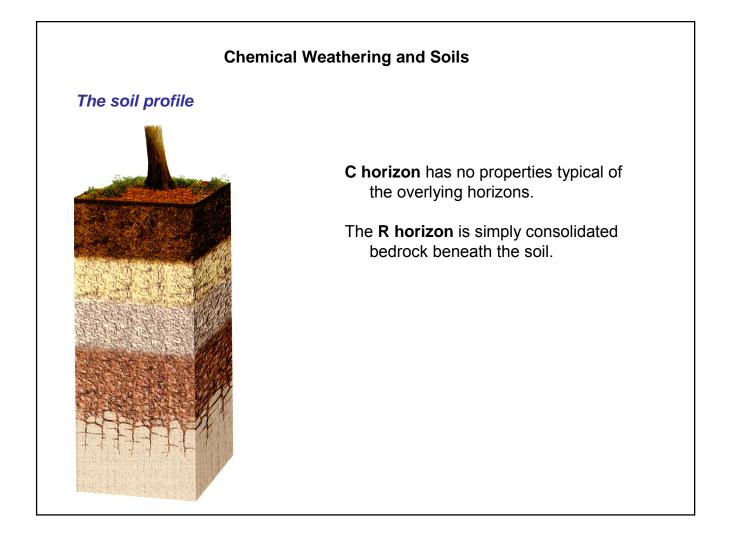


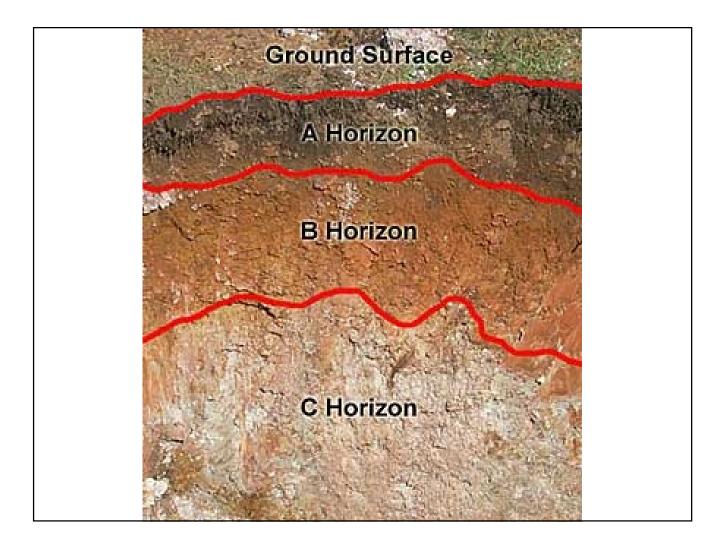


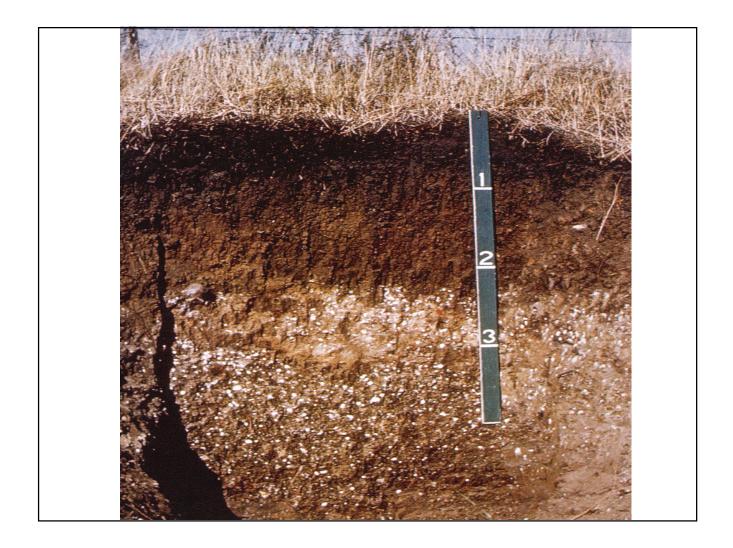




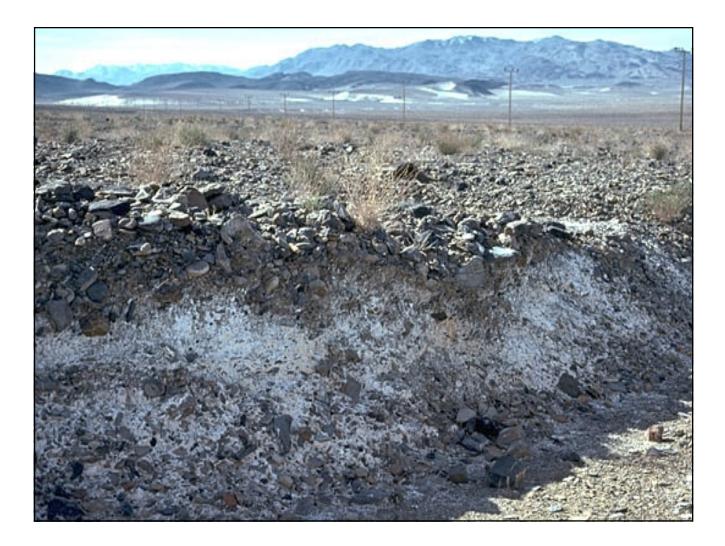






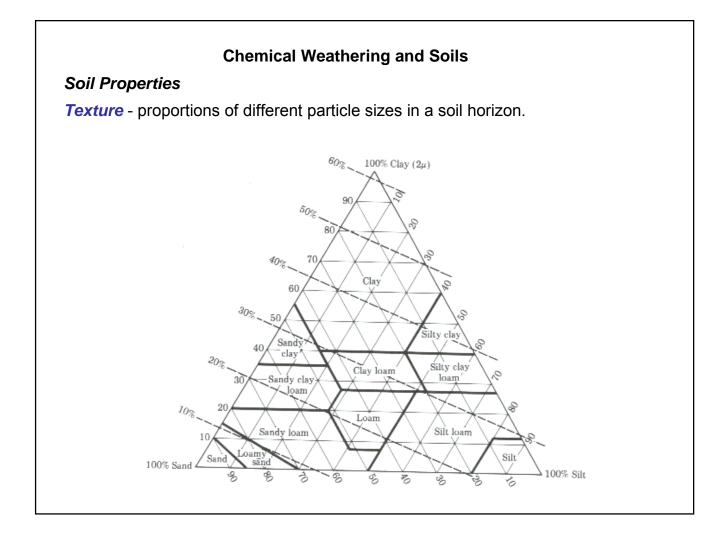


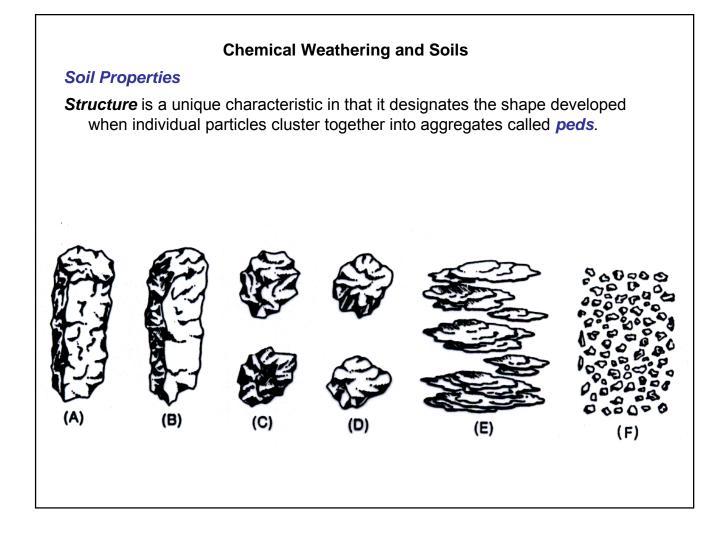




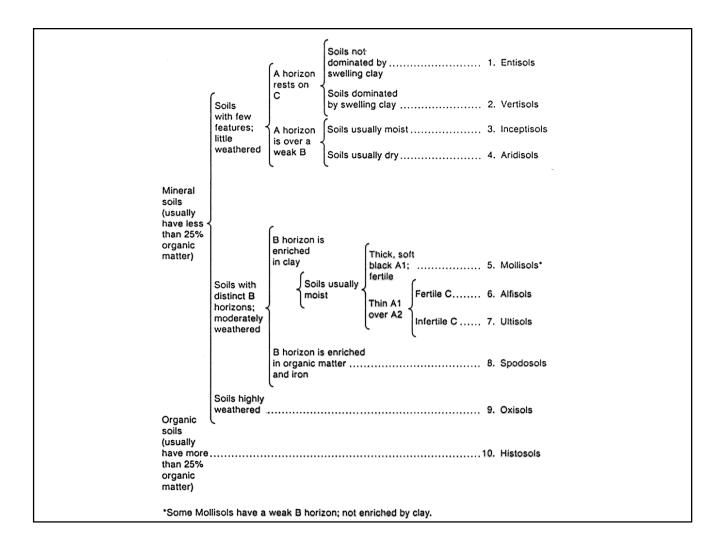


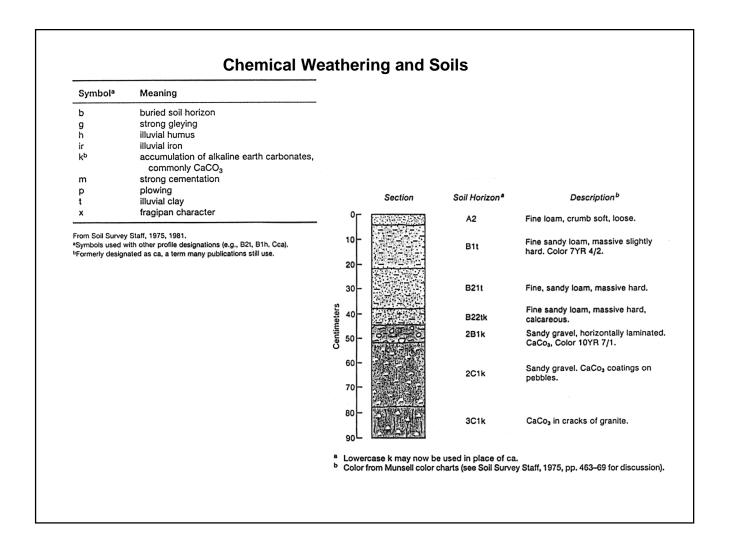


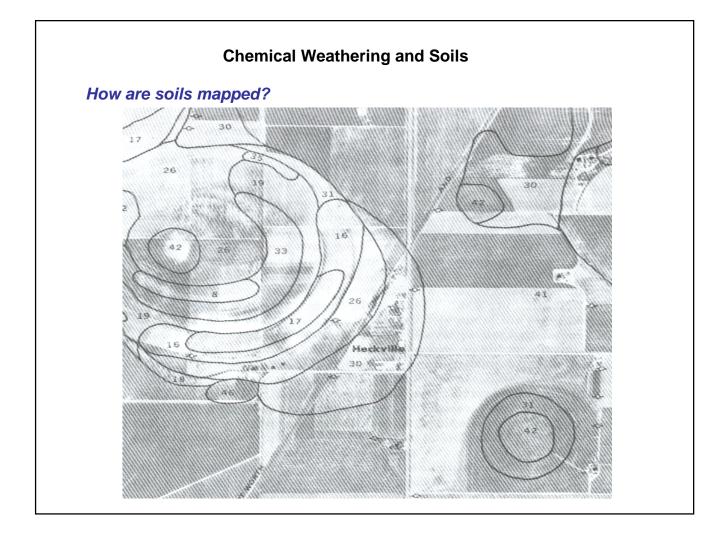


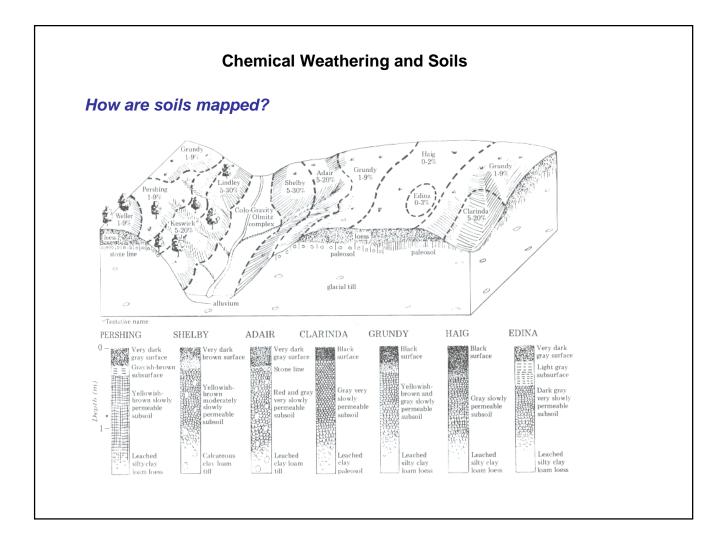


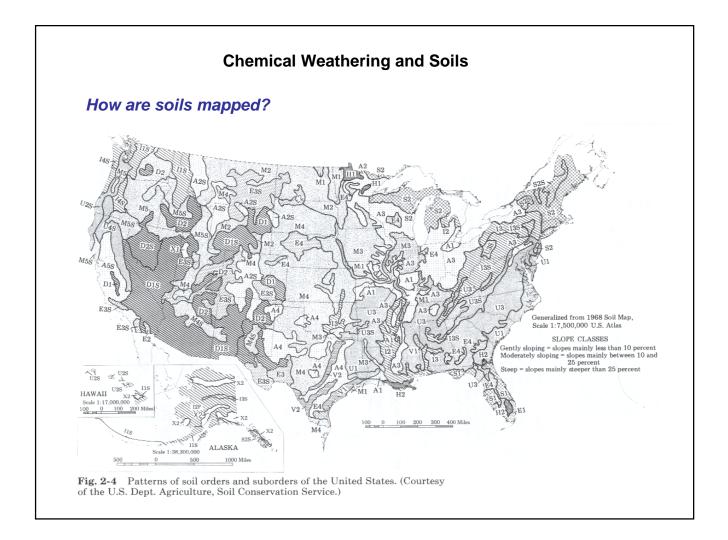
Chemic Soil classification major orders are		eathering and Soils
Entisols	-	recent
Vertisols	-	inverts
Inceptisols	-	inception
Aridisols	-	arid
Mollisols	-	mollify (to lessen in intensity, to soften
Spodosol	-	podzol, odd
Alfisol	-	pedalfer
Ultisol	-	ultimate
Oxisol	-	oxide
Histosol	-	histology (tissue)



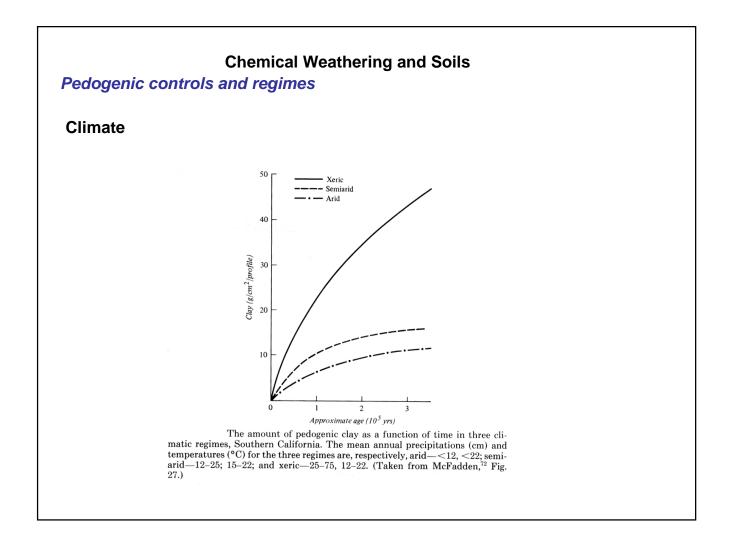


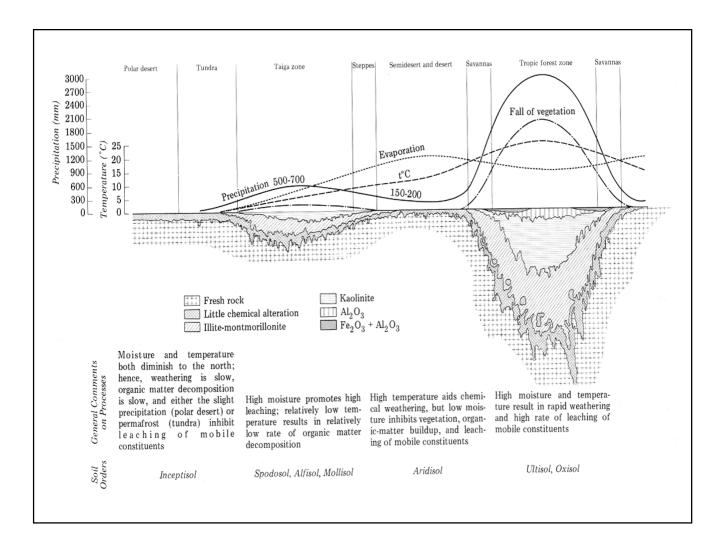


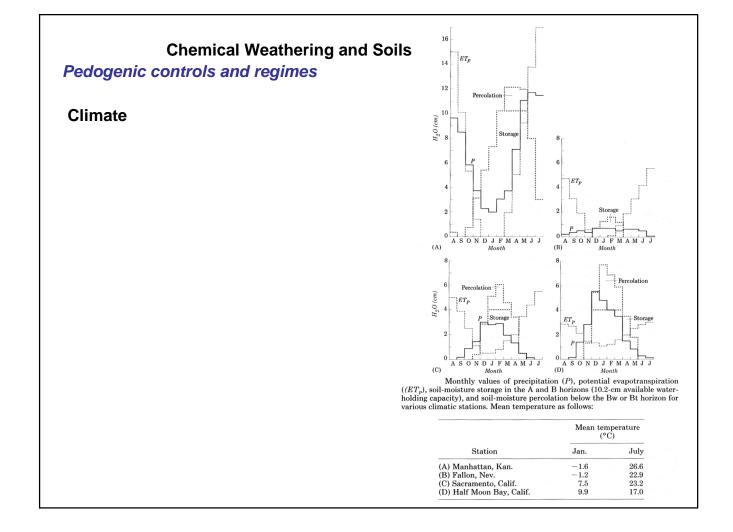


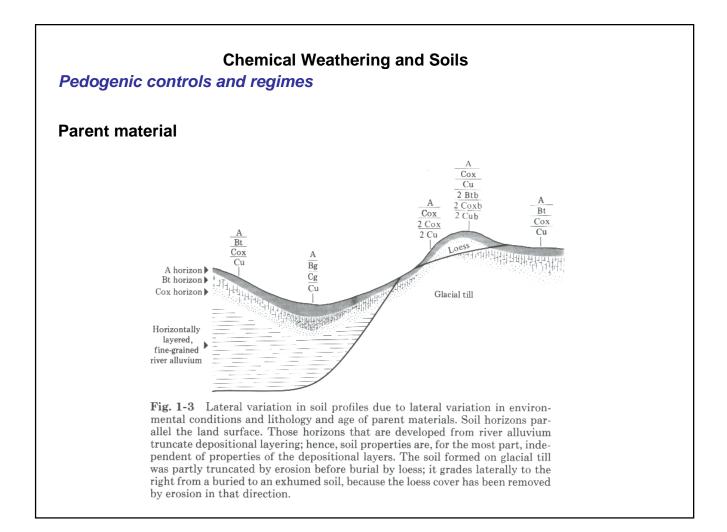


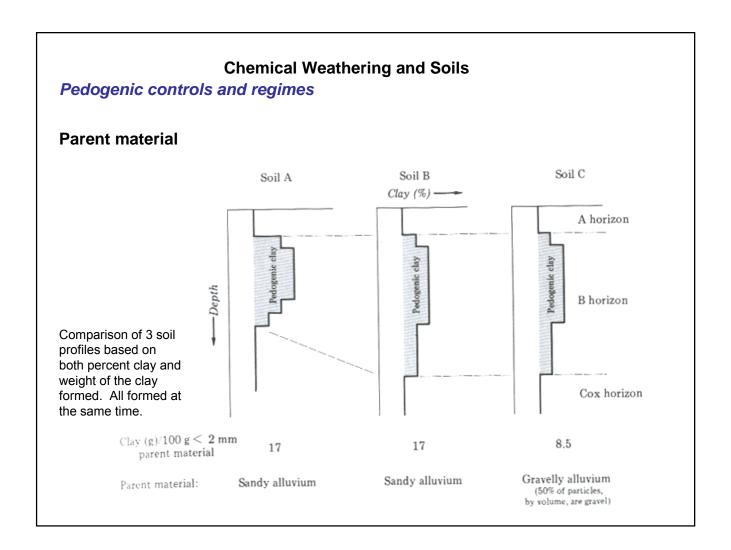
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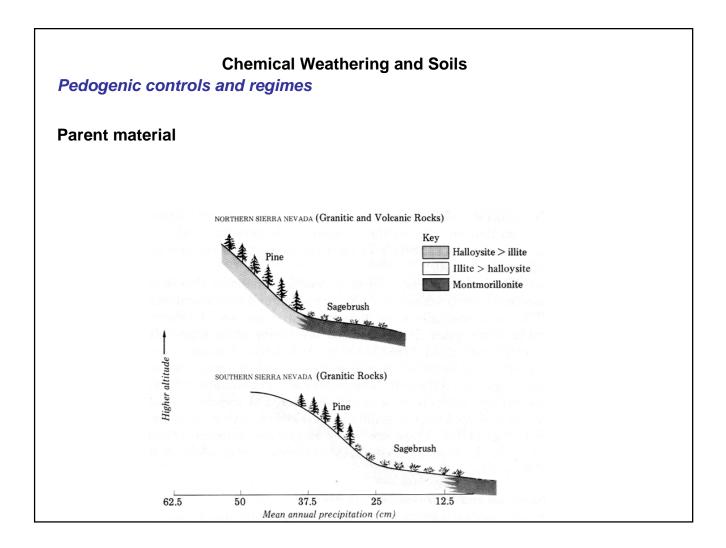


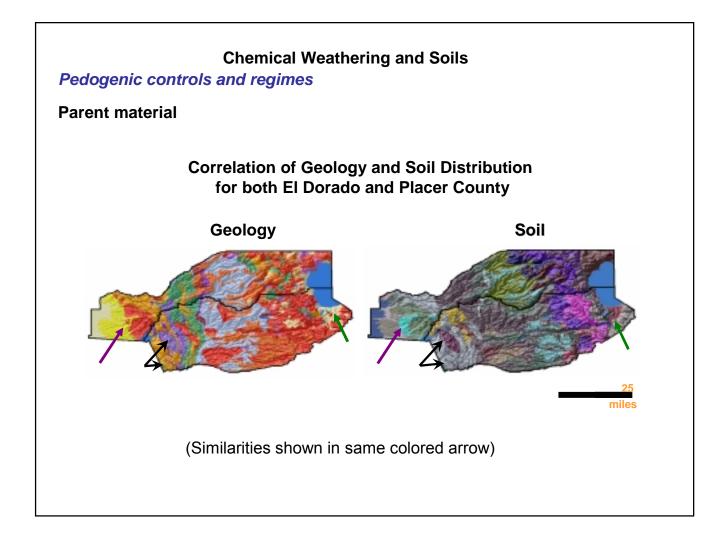


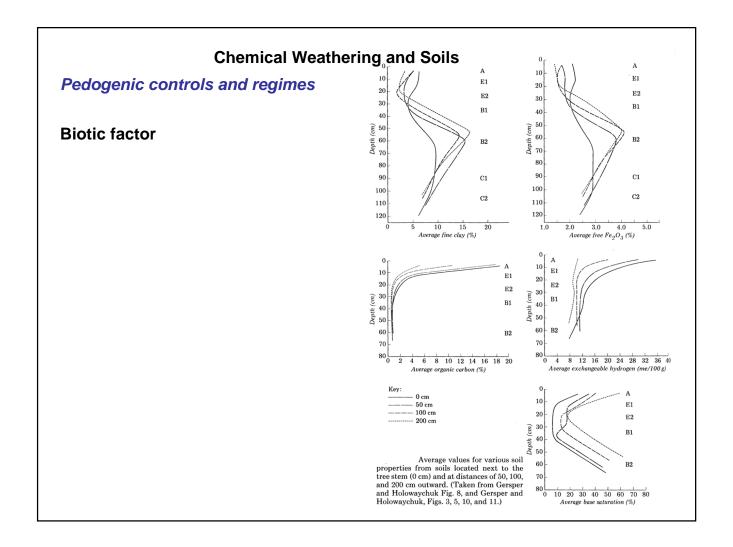


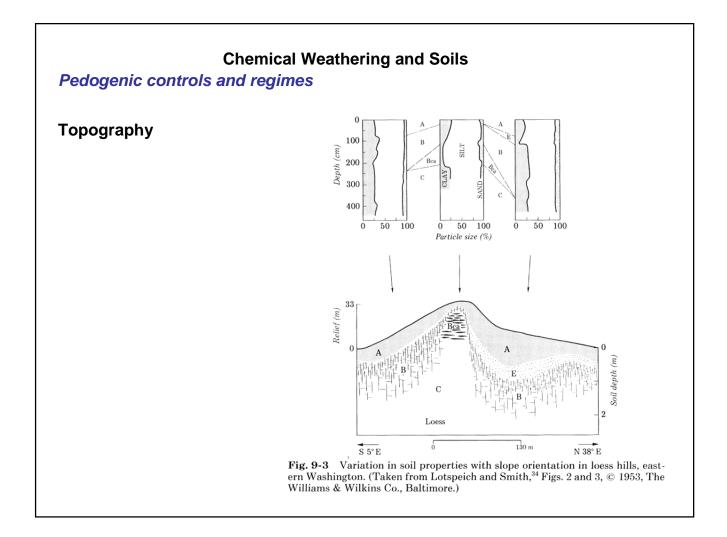


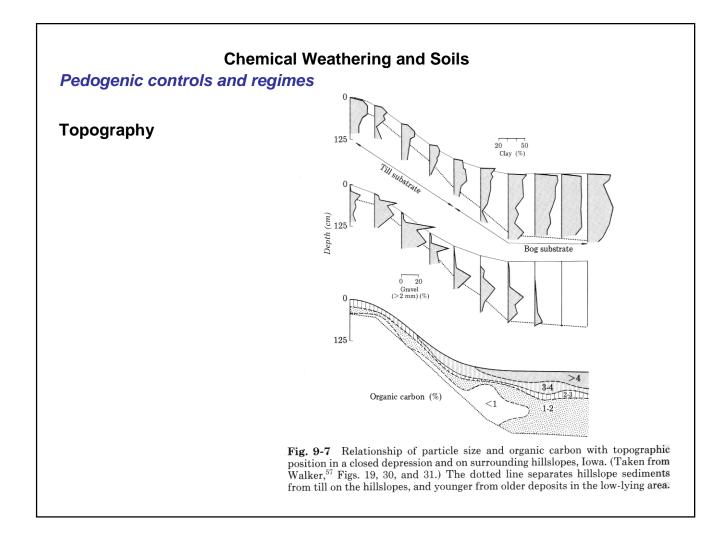


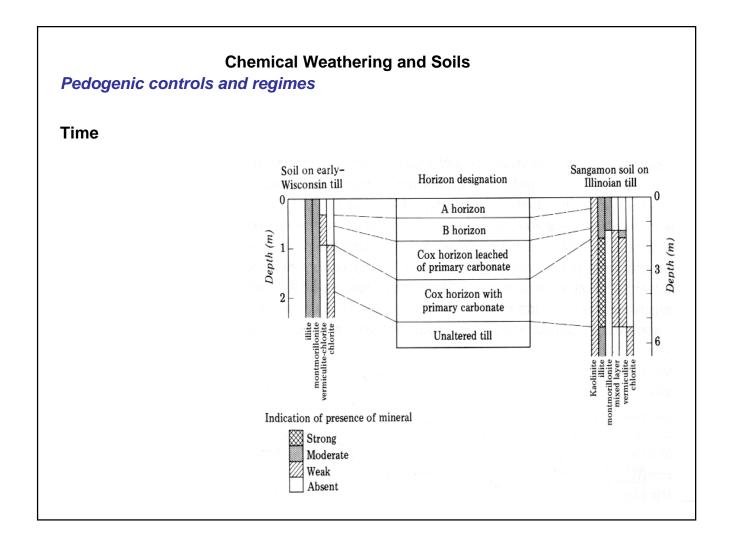


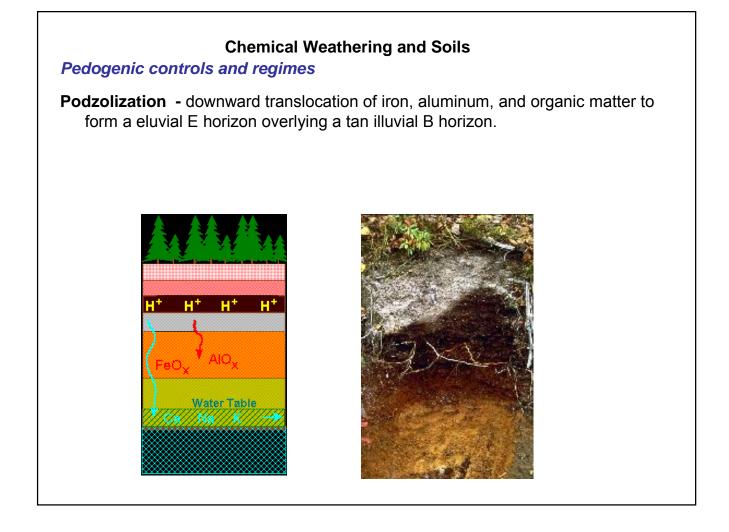


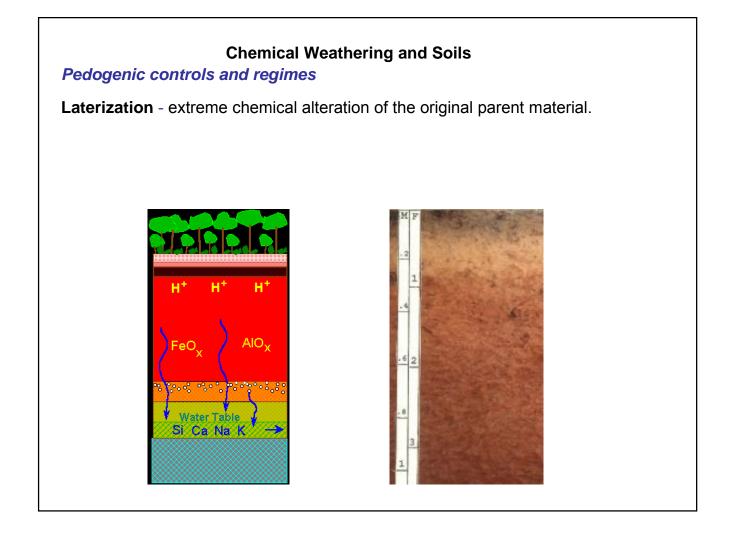












Pedogenic controls and regimes

Calcification - a general buildup of calcium carbonate in the soil profile.



Pedogenic controls and regimes

Salinization - soluble salts precipitate from water and accumulate in the soil. Saline soils are common in desert and steppe climates.



Geomorphic significance of soils

They are used primarilyin the subdivision of a local succession of deposits,



Geomorphic significance of soils

They are used primarily

- to provide data on the lengths of time that separate periods of deposition,
- to facilitate short- and long-term correlation, and, of recent discovery,
- to date Quaternary events (faulting, climate change, etc.)
- archeological studies
- Reconstruct climates



